**Department:** Science

Course Name: General Chemistry

# **Course Description:**

The General Chemistry course is designed to focus on the basic principles of chemistry that includes a study of elements, structure of matter, and bonding fundamentals. This course incorporates a conceptual approach to chemistry with less emphasis placed on problem solving involving mathematical applications. Regular laboratory work is an integral part of this course.

## **Content:**

Introduction to Chemistry

Matter and Change

Physical and Chemical Properties

Scientific Measurement

Significant Figures

Metric Units

**Dimensional Analysis** 

Structure of the Atom

Electronic Configuration of Atom

Properties of the Periodic Table

Ionic, Metallic, Ionic & Covalent Bonding

Molecular Geometry

Chemical Names & Formulas

Chemical Quantities, Moles

**Chemical Reactions** 

Stoichiometry

Behavior of Gases

Acids, Bases, and Salts

**Organic Chemistry** 

**Nuclear Chemistry** 

## **Skills:**

Apply math concepts to the chemistry

Gather and analyze data

Construct lab reports

Develop basic computer skills

Apply reason, logic, and basic problem-solving skills

#### **Text and Materials:**

Wilbraham, Staley, Matta, Waterman, Chemistry, (Pearson, 2012)

#### **Methods of Instruction:**

Lecture

Laboratory experiments

**Demonstrations** 

Computer projects

Class projects

Collaborative electronic documents

Instructional videos Worksheets **Methods of Evaluation:** Quizzes Tests Homework Class work Class participation Laboratory reports Laboratory participation